

TOUGH Short Course

Lawrence Berkeley National Laboratory
Earth Sciences Division
Berkeley, California

Radionuclide Transport for TOUGH2

- Description of EOS7R
- Physical Processes
- Primary Variables
- Input Formats

1

EOS7R

- EOS7R is an equation of state module for simulation with:
 - Water, brine, air, 2 radionuclides, and heat
 - Radionuclides that decay and are water soluble and volatile
 - Adsorption onto solid grains
 - Molecular diffusion in aqueous and gas phases
- Assumptions
 - Radionuclides do not form non-aqueous phases
 - Salt precipitation does not occur
- Additional reference:
 - Oldenburg and Pruess, EOS7R: *Radionuclide Transport for TOUGH2*, Report LBL-34868, LBNL, Berkeley, Calif., Nov., 1995.
- Code executable created by compiling TOUGH2 with *EOS7R.f*. Additional instructions in *read.me* file.

2

Radioactive Decay

- Exponential decay of mass (M) of radionuclide component (k)

$$\begin{aligned}\frac{dM^k}{dt} &= -\lambda_k M^k \\ M^k &= M_0^k \exp(-\lambda_k t)\end{aligned}$$

- Half-life depends on decay constant λ_k

$$\begin{aligned}\frac{M_{t_{1/2}}^k}{M_{t_0}^k} &= \exp(-\lambda_k t_{1/2}) = \frac{1}{2} \\ t_{1/2} &= \frac{\ln 2}{\lambda_k}\end{aligned}$$

3

Sorption and Dissolution of Radionuclides

- Mass adsorbed onto solid grains by reversible instantaneous linear model

$$M^k = \phi \sum_{\beta} S_{\beta} \rho_{\beta} X_{\beta}^k + (1 - \phi) \rho_R \rho_{aq} X_{aq}^k K_d$$

- Gas dissolution in the aqueous phase determined by Henry's law

$$P_k = K_h X_{aq}^k \quad \text{or} \quad X_{aq}^k = (K_h^{-1}) P_k$$

The amount of component k dissolved in aqueous phase is determined by the partial pressure and the “inverse of Henry's constant”

4

Primary Variables

- EOS7R considers up to a five components plus heat
-Water, brine, parent radionuclide (Rn1), daughter radionuclide, air, and heat
- The possible primary variables are:

Pressure,	P
Mass fraction of brine,	X_b
Mass fraction of Rn1,	X_{Rn1}
Mass fraction of Rn2,	X_{Rn2}
Mass fraction of air,	X_{air}
Gas phase saturation,	S_g
Temperature,	T

5

Primary Variables

- Two combinations of primary variables (X1-X6) are possible, depending on phase condition:

	X1	X2	X3	X4	X5	X6
1-phase (aq.)	P	X_b	X_{Rn1}	X_{Rn2}	X_{air}	T
2-phase (aq. + gas)	P	X_b	X_{Rn1}	X_{Rn2}	S_g+10	T

- When $P_{gas} > P_{aq}$, transition to 2-phase from 1-phase conditions
- When $S_g < 1$, transition to 1-phase from 2-phase conditions
- Number of unknowns can be reduced in MULTI block (see p.45, 162)
 - If gas is excluded, NK = 4 (NK = 5 otherwise)
 - If diffusion is ignored, NB = 6 (NB = 8 otherwise)

6

Input Format for EOS7R

- Properties related to diffusion, adsorption, radioactive decay and volatilization are specified in block SELEC (see p. 46 of Manual)

TOUGH2 INPUT FORMATS (continued)

INDOM	(optional) 1	2	3	4	5	6	7	8
MAT								
	X1	X2	X3	X4				
DIFFU	(optional) 1	2	3	4	5	6	7	8
	FDDIAC(I,1), I=1, NPH							
	FDDIAC(I,2), I=1, NPH							
SELEC	(optional) 1	2	3	4	5	6	7	8
	IE(1)	IE(2)	IE(3)	IE(4)	IE(5)	IE(6)	IE(7)	IE(8)
	FE(1)	FE(2)	FE(3)	FE(4)	FE(5)	FE(6)	FE(7)	FE(8)
	FE(9)	FE(10)	FE(11)	FE(12)	FE(13)	FE(14)	FE(15)	FE(16)
	FE(17)							
								FE(8*IE(1))

SELEC.1
 SELEC.2
 SELEC.3
 SELEC.4
 ...